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NEW COUMARINS FROM THE AILANTHUS ALTISSIMA

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Abstract – Two new coumarin derivatives were isolated together with three known coumarins, artelin (3), isofraxidin (4), and scoploetin (5) from the bark of *Ailanthus altissima* (Simaroubaceae). New coumarin derivatives were elucidated as terpenylated coumarins, named altissimacoumarin A (1) and altissimacoumarin B (2), respectively. *Trans*-Configuration of between C-2' and C-9' in compound (1) was clearly confirmed by NOESY experiments.

Ailanthus altissima is classified in the subfamily Simaroubaceae, which has been used for a treatment of colds and gastric diseases in Chinese traditional medicine.^{1,2} Previous researches revealed that *A. altissima* has many pharmacologically active constituents such as quassinoids,^{3,4} alkaloids⁵ and lipids.⁶ In continuing chemical studies on medicinal plants, we investigated the chemical constituents from this species. A methanol extract of the bark of *A. altissima* gave two new terpenylated coumarins, named altissimacoumarin A (1) and altissimacoumarin B (2) as well as three known coumarins, artelin (3), isofraxidin (4), and scoploetin (5).



The known compounds, artelin (3), isofraxidin (4), and scoploetin (5) were identified by comparison of their spectral data with literature data.^{7,8} The molecular formula of compound (1) was established as $C_{21}H_{26}O_7$ by HRMS (m/z = 390.1677) and DEPT results of ¹³C-NMR spectra. Base peak in MS fragment at m/z 222 [M-168]⁺ suggested the presence of a isofraxidin moiety. The lactonic group in coumarin skeleton was confirmed with a absorption at 1730 cm⁻¹ in IR spectrum. In the ¹H-NMR spectrum, the splitting patterns of a pair of doublets [δ 6.35 and 7.62 (each 1H, d, J = 9.5 Hz)], a one-proton singlet at δ 6.68 (1H, s), and two methoxy signals [\delta 3.89 and 4.04 (each 3H, s)] were consistent with those of the isolated isofraxidin (4). Hence, the extra ten carbons in ¹³C-NMR spectrum were deduced the signals from terpenyl group. The terpenyl group was confirmed with ¹H-¹H COSY spectrum which revealed successive connectivities from C-1' to C-2' and from C-4' to C-6'. Unassigned connectivities of terpenyl group, C-4'/C-9' and C-2'/C-9', were determined on the basis of HMBC correlations (Figure 2). The proton at δ 3.24 and carbon at δ 60.5 suggested the presence of oxiran ring. Finally, the tertiary hydroxy group was confirmed with a absorption at 3450 cm⁻¹ in IR spectrum. This oxygenated terpenyl group was attached to the C-7 position because the H-1' proton resonating at δ 4.26 displayed HMBC connectivity with C-7. Thus, compound (1) was clearly defined as 7-(3',7'-dimethyl-7-hydroxy-2',3'-oxy-5octenyl)oxyisofraxidin, named altissimacoumarin A. The trans configurations of the H-2' and H-9' were determined by 2D-NOESY experiments. NOE cross peaks were observed between H-2' and H-4', whereas NOE was not at H-2' and H-9'. Moreover, strong NOE cross peaks were observed between H-1' and H-9' (Figure 1).



Figure 1 Selected NOE correlations of 1

Compound (2) was found to have the molecular formula of $C_{21}H_{26}O_6$ from HRMS (m/z = 374.1730) and DEPT results of ¹³C-NMR spectra. Base peak in MS fragment at m/z 222 [M-152]⁺ suggested the presence of a isofraxidin moiety which showed same pattern of ¹H- and ¹³C-NMR data with compound 1. The terpenyl group was also determined on the basis of ¹H-¹H COSY and HMBC correlations (Figure 2). The proton at δ 3.21 and carbon at δ 60.8 suggested the presence of oxiran ring in terpenyl group. This oxygenated terpenyl group was attached to the C-7 position because the H-1' proton resonating at δ 4.25 displayed HMBC connectivity with C-7. Thus, compound (2) was clearly defined as 7-(3',7'-dimethyl-2',3'-epoxygeranyl)oxyisofraxidin, named altissimacoumarin B; it showed optical activity with [α]_D²⁰+18.2° (*c*, 0.50, CHCl₃).



Figure 2 Important HMBC correlations of 1 and 2

Position	1 2			
	$^{1}\mathrm{H}$	¹³ C	$^{1}\mathrm{H}$	¹³ C
2	-	160.7 s	-	160.7 s
3	6.35 (1H, d, 9.5)	115.9 d	6.34 (1H, d, 9.5)	115.8 d
4	7.62 (1H, d, 9.5)	143.7 d	7.66 (1H, d, 9.5)	143.7 d
5	6.68 (1H, s)	104.3 d	6.67 (1H, s)	104.2 d
6	-	150.6 s	-	150.6 s
7	-	143.0 s	-	145.1 s
8	-	141.9 s	-	141.9 s
9	-	115.2 s	-	115.1 s
10	-	143.3 s	-	143.3 s
1'	4.26 (2H, dd, 2.1, 5.6)	72.6 t	4.25 (2H, m)	72.9 t
2'	3.24 (1H, t, 5.6)	60.0 d	3.21 (1H, t, 6.1)	60.9 d
3'	-	60.5 s	-	60.8 s
4'	2.34 (2H, m)	41.1 t	1.66 (1H, m)	38.7 t
			1.53 (1H, m)	
5'	5.65 (1H, m)	125.7 d	2.08 (2H, dd, 12.9, 25.7)	23.9 t
6'	5.65 (1H, m)	137.9 d	5.10 (1H, m)	123.8 d
7'	-	82.3 s	-	132.4 s
8'	1.32 (3H, s)	24.5 q	1.61 (3H, s)	18.0 q
9'	1.26 (3H, s)	17.4 q	1.25 (3H, s)	17.1 q
10'	1.32 (3H, s)	24.6 q	1.68 (3H, s)	26.0 q
6-OMe	3.89 (3H, s)	56.8 q	3.90 (3H, s)	56.7 q
8-OMe	4.04 (3H, s)	62.2 q	4.05 (3H, s)	62.2 q

Table 1. The ¹H- and ¹³C-NMR spectral data of **1** and **2** (CDCl₃)

EXPERIMENTAL

Plant materials. Bark materials of *Ailanthus altissima* Swingle were collected in Jinju (Korea) and identified by Prof. Myong Gi Chung. A voucher specimen (*S. W. Hwang & M. S. Yang 022*) of this raw material is deposited at Herbarium of the Gyeongsang National University (GSNU).

General experimental procedures. Optical rotations were obtained using a Perkin-Elmer polarimeter. IR spectra were recorded on a Bruker IFS66 and UV spectra were measured on a Beckman DU650 spectrophotometer. ¹H and ¹³C-NMR spectra along with 2D-NMR spectral data were obtained on a Bruker AM 500 (¹H-NMR at 500 MHz, ¹³C-NMR at 125 MHz) spectrometer in CDCl₃ solution. EIMS and HREIMS spectra were recorded on a JEOL JMS-700 instrument operated at 70eV.

Extraction and Isolation. The air-dried bark (2 kg) of *A. altissima* was extracted with MeOH (10 $L \times 3$) at rt for 72 h. The combined extract was concentrated *in vacuo* to afford a brown gum (120 g), which was partitioned with chloroform and water. The chloroform layer was washed with brine, dried over Na₂SO₄, and then concentrated to give a thickish residue (36 g). The residue was chromatographed

on a silica gel (500 g) column eluted with a gradient of 100% chloroform to 100% MeOH to afford 72 fractions (F1-F72, each 250 mL). F 25-32 were combined and applied to a silica gel column, eluted with hexane-ethyl acetate mixtures of increasing polarity (49 : 1 \rightarrow 1 : 1, each 50 mL), to give 40 subfractions (A1-A40). Fractions A21-A25 were further purified with silica gel chromatography eluting with hexane and ethyl acetate (9 : 1 \rightarrow 1 : 1) to afford compound (1) (10 mg, R_f =0.32, hexane-ethyl acetate = 1 : 1), **3** (18 mg, R_f =0.71, hexane-ethyl acetate = 1 : 1), and **4** (15 mg, R_f =0.45, hexane-ethyl acetate = 1 : 1). Fractions A26-A29 were further purified with silica gel chromatography eluting with chloroform and acetone (49 : 1 \rightarrow 1 : 1) to afford compound (**2**) (28 mg, R_f =0.75, chloroform-acetone = 9 : 1) and **5** (15 mg, R_f =0.42, chloroform-acetone = 9:1).

Altissimacoumarin A (1): $[\alpha]_D^{20}$ +11.1° (*c*, 0.50, CHCl₃); UV λ_{max} : 223, 293, 333, 339 nm (MeOH); IR ν_{max} (KBR)cm⁻¹: 3436, 1732 nm; EIMS: $m/z = 390 [M]^+$ (4.8), 372 (7.5), 222 (100), 207 (14), 179 (7.5), 150 (11.2), 107 (22.1); HREIMS: m/z = 390.1677 (calcd for C₂₁H₂₆O₇ 390.1679); ¹H- and ¹³C-NMR (see Table 1).

Altissimacoumarin B (2): $[\alpha]_D^{20}$ +18.2° (*c*, 0.50, CHCl₃); UV λ_{max} : 225, 293, 339, 395 nm (MeOH); IR v_{max} (KBR)cm⁻¹: 1732, 1565, 1458 nm; EIMS: $m/z = 374 [M]^+$ (7.5), 252 (9.2), 222 (100), 221 (10.2), 207 (9.9), 150 (5.2), 135(10.2); HREIMS: m/z = 374.1730 (calcd for C₂₁H₂₆O₆ 374.1729); ¹H- and ¹³C-NMR (see Table 1).

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REFERENCES

- S. K. Jain and R. A. DeFilipps, 'Medicinal Plants of India', Reference Publications Inc., Michigan, 1991.
- 2. D. M. Vincenzo, D. M. Laura, Q. Emilia, and P. Cosimo, J. Agr. Food Chem., 2003, 51, 1177.
- S. Tamura, N. Fukamiya, M. Okano, J. Koyama, K. Koike, H. Tokuda, W. Aoi, J. Takayasu, M. Kuchide, and H. Nishino, *Chem. Pharm. Bull.*, 2003, 51, 385.
- K. Kubota, N. Fukamiya, T. Hamada, M. Okano, K. Tagahara, and K. H. Lee, *J. Nat. Prod.*, 1996, 59, 683.
- 5. T. Ohmoto, K. Koike, and Y. Sakamoto, Chem. Pharm. Bull., 1981, 29, 390.
- 6. G. Bory and D. Chair-Maezulajtys, Indian J. Plant Physiol., 1989, 32, 12.
- 7. O. Kayser and H. Kolodziej, *Phytochemistry*, 1995, **39**, 1181.
- 8. D. V. Banthorpe and G. D. Brown, *Phytochemistry*, 1989, 28, 3003.